



Thorium as a nuclear material: physics features and a safety analysis due to insertion of ThO2 in a PWR nuclear reactor fuel

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ABSTRACT

On this paper it is presented the main physics features of the thorium as a nuclear material and a security analysis due to insertion of ThO2 in a PWR nuclear reactor fuel by the determination and assessment of the sensitivity coefficients through the Standardized Computer Analyses for Licensing Evaluation - SCALE 6.1 nuclear code system. This work was motivated by the world tendencies to research and development of nuclear fuels based on thorium, because the limitations of uranium resources and the importance of nuclear energy to the humanity.

Keywords: Thorium Dioxide, Physics Features of Thorium, Thorium As a Nuclear Material.

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1. INTRODU'CTION

The human development it is closely related to the level and type of power supply by us dominated. At the beginning of the last century it was discovered and controlled the nuclear fission energy the most productive source of energy available, anytime, anywhere.

The progress of nuclear energy research and development of the study about the exploitation of thorium as another nuclear material, besides the uranium, was initialized at the Oak Ridge National Laboratories - ORNL on EUA at mid 50's years [1]. In Brazil, the first approach to the researchs about thorium as a nuclear material occurred during the commercial agreement between Brazil and France at the mid of the 60's years and last, with the project codenamed Toruna, until the end of 60's years and the 70's beginning. The research continued under the cooperative agreement between Brazil and Germany [2]. The researches about thorium were interrupted until the 90's years, when many countries resumed the research and development about the thorium utilization in nuclear power plants.

Currently there are 448 nuclear power reactors in operation and 59 under construction. Of the reactors in operation: 292 are of the Pressurized Water Reactor (PWR) type; 75 of the Boiling Water Reactor (BWR) type; 14 Gas Cooled Reactor (GCR); 49 Pressurized Heavy Water Reactor (PHWR); 15 Light Water Graphite Reactor (LWGR) and 3 of the Feast Breeder Reactor (FBR) type. Already under construction: 49 are PWR; 4 BWR; 4 PHWR; 1 FBR and 1 Hight Temperature Gas Cooled Reactor (HTGR), according to International atomic Energy Agency - IAEA [3]. In 2017, nuclear energy represented 11% of the world's energy matrix. In an extrapolated perspective, this percentage would be 12.4% in 2030, 13.4% in 2040 and 13.7% in 2050 [4].

Most power reactors, in operation or under construction, use uranium as fuel. With the implementation of new reactors and the increase in energy demand, especially clean energy generation, the demand for uranium is expected to continue to grow. With the 2017 reserve estimates and the level of demand for U in 2016, it is assumed that uranium resources will be sufficient to supply its global demand for another 130 years according to Organization for Economic Co-operation and Development - OECD [5].

The need for nuclear power coupled with the perspective of a horizon for uranium reserves leads to research and exploration of new resources. The most direct solution is thorium, which in addition to having facilitated exploration, due to the intrinsic characteristics of its ore, mainly thorium, is more abundant than uranium in the earth's crust and also presents complicating factors for proliferation. In Brazil, thorium natural reserves are among the largest in the world, IAEA – Nuclear Energy Agency - NEA [6].

1.1. Torium as nuclear material

Thorium (Th) is the first nuclear material of natural occurrence, besides followed by uranium, and its natural abundance is on average 3.4 times greater than that of U. Its most recurrent isotope is ${}^{232}_{90}Th$, a fertile isotope that when fertilized by a neutron, in a reaction of neutron-nuclear absorption, is transmuted into the fissile isotope o ${}^{233}_{92}U$ [7].

An illustration of the transmutation process of the $^{232}_{90}Th$ to the $^{233}_{92}U$ can be seen at the Figure 1.

Figure 1: Simplified fertilization process of the ²³²Th to fission of the ²³³U.



Source: Adapted, [8].

Thorium dioxide is one of the most chemically stable solid substances, its melting point is one of the highest among the known, 2,023K - elementary and 3,643K - dioxide, specific mass 9,86g/cc, resistant to oxidation, thermal conductivity 43W/(m.K) - elementary or (6.2 W)/(m.K) at 773K - dioxide and 2.4W/(m.K) at 1773 K – dioxide [9] [10].

In order to perform a comparative analysis, Table 1 presents the neutronic properties of the main nuclear materials.

| Nuclear s Data | ²³² Th | ²³³ U | ²³⁵ U | ²³⁸ U | ²³⁹ Pu | ²⁴¹ Pu |
|----------------------------------|-------------------|------------------|------------------|------------------|-------------------|-------------------|
| Thermal Region | | | | | | |
| Cross Sections | | | | | | |
| [barns] | | | | | | |
| Absortion σ_a | 4.62 | 364 | 405 | 1.73 | 1,045 | 1,121 |
| Fission σ_f | 0 | 332 | 346 | 0 | 695.0 | 842 |
| $\alpha = \sigma_c / \sigma_f$ | - | 0.096 | 0.171 | | 0.504 | 0.331 |
| η_{th} | - | 2.26 | 2.08 | | 1.81 | 2.23 |
| Region of Resonance | | | | | | |
| Integral Epitermal (RI) [barns] | 0 | 764 | 275 | 0 | 301 | - |
| $(\propto dilution)$ | | | | | | |
| <i>RI_a</i> | 85.6 | 882 | 405 | 278 | 474 | 740 |
| $_$ RI_{f} | - | 746 | 272 | | 293 | 571 |
| $\alpha = RI_a/RI_f$ | - | 0.182 | 0.489 | | 0.618 | 0.296 |
| η_{epi} | - | 2.10 | 1.63 | | 1.77 | 2.29 |
| Neutronic Yield ν | - | 2.43 | 2.43 | | 2.87 | 2.97 |
| Delay Neutronic Yield β | - | 0.0031 | 0.0069 | | 0.0026 | 0.0050 |
| Capture: | | | | | | |
| 2.200 m/s | 7.6 | 54 | 100 | 2.7 | 267 | - |
| Integral Resonance | 85 | 140 | 144 | 275 | 200 | - |
| Average neutrons by fission | | 2.5 | 2.4 | - | - | 2.9 |

Table 1: Neutronic properties of the main fissile and fertile isotopes.

Source: Adapted, [7]

Analyzing Table 1 it can be verified that the absorption cross section of ^{232}Th is two times greater than that of ^{238}U indicating that in regenerative reactors the probability of conversion of ^{232}Th to ^{233}U is greater than ^{238}U to ^{239}Pu . The behaviour of the ^{232}Th and ^{238}U neutron capture cross sections with respect to the energy of the incident neutron is shown in Figure 2.

Figure 2: Behaviour of the .²³²Th and ²³⁸U neutron Capture Cross sections.



For comparative purposes, the behavior of the fissile isotope neutron capture cross sections ^{233}U , ^{235}U and ^{239}Pu is presented in Figure 3.



Figure 3: Cross section spectrum for the isotopes ²³³U, ²³⁵U and ²³⁹Pu.

One approach to determine how to a perturbation or change in some specific parameter or feature of the nuclear reactor affects the stead state of its operation, for example the response of the neutron multiplication factor due to a change in the coolant/moderator temperature of a PWR nuclear reactor, is to evaluate the so called sensitivity coefficients. According to this, the well stablished linear perturbation theory can be used to determine the effects of a perturbation in the reactivity and, therefore in the safety of a nuclear reactor due to that perturbation, [11]. A simplified theoretical approach to the linear perturbation theory is presented by DUDERSTADT (1976) [12].

2. MATERIALS AND METHODS

To perform the sensitivity analysis of the neutron multiplication factor, in this work the infinite neutron multiplication factor - k_{inf} , it was used the nuclears code system SCALE 6.1. One of its control modules, the TSUNAMI – Tools for Sensitivity and UNcertainty Methodology Implementation, accomplishes the sensitivity analysis.

TSUNAMI modules, by allowing the analysis of a system intrinsically, performs an internal computational analysis with its own method. However, it is possible to repeat the simulations with changes in some of the data parameters of the input file allowing an analysis of the influence of this change in the sensitivity and uncertainty due to the variable object of the study [8]. The extrinsic factor chosen to be varied in this work is the temperature of the refrigerant, maintaining constant the temperatures of the other components of the fuel, and it was performed the analysis of the behavior of the k_{inf} , the sensitivity and the uncertainty for all the materials and regions and fuel due to the change.

For this work just the sensitivity will be considered in the analysis, disregarding the fuel burn-up. It was chosen the nuclear power plant data of Angra II, wich has three levels of fuel enrichment, in terms of the isotopic concentration of ^{235}U , 1.95%, 2.5% and 3.2% and for this work was take just the 2.5% and 3.2% of levels of enrichment, considering the first FSAR – "Final Safety Analyses Report" [13].

As the fuel to be analyzed is a mix of UO_2 and ThO_2 it was verified that it should be necessary to calculate the relative quantities of each elements from the fuel mix, as presented on Tables 2 and 3 at the Chapter 3 of this paper results. To achieve this goal it was selected an approach by stoichiometric balance, which one was done by the following equations, where the parameter '*x*' is the weight percent (%wt) of ThO_2 at the fuel mix, '*e*' is the enrichment of uranium and '*M*' is the molar mass of the dioxides.

$$M_{tot} = x. M_{ThO_2} + (1 - x). M_{UO_2}.$$
 (1)

$$x. M_{ThO_2} = x. M_{Th} + .2 x. (0.997. M_{0^{16}} + 0.003. M_{0^{17}})$$
(2)

$$(1-x). M_{UO_2} = (1-x). e. M_{U^{235}} + (1-x). (1-e). M_{U^{238}} + (1-x). 2. (0,997. M_{O^{16}} + 0,003. M_{O^{17}})$$
(3)

At the end of the Eq. (3) it was specified the percent of each isotopes of the fuel mix.

To calculate the density of the different full mix was choose a weighted averaged by the % wt ThO_2 .

Fonseca et al. • Braz. J. Rad. Sci. • 2020

$$\rho \left[{g \atop cm^3} \right] = x. \rho_{ThO_2} + (1-x). \rho_{UO_2}$$
(4)

7

Considering the input data of the fuel composition it was chosen the temperature range from 280K to 345K to perform the simulations, and the range of %wt of ThO₂ from 0%wt to 40% wt. The SAMS, the SCALE functional modules to evaluate the sensitivity and uncertainty, as a subroutine of TSUNAMI, generate two different sensitivity coefficients: one explicit, due to the perturbations inserted, and the other came from the statistical analyses of the error propagation on the calculation of the k_{inf} . The sum of the two coefficients results in the total sensitivity coefficients estimated for each region characterized material or mixture. These results are presented in the form of graphics and tables on Chapter 3.

3. RESULTS AND DISCUSSION

Tables 2 and 3 below presents the composition of the fuel mix to both levels of enrichment including different percent mass of ThO2.

| %ThO2 Comp. | 0 | 5 | 10 | 15 | 20 | 25 | 30 | 35 | 40 |
|-----------------------|--------|--------|--------|--------|--------|--------|--------|--------|--------|
| Total Mass [g] | 269.93 | 269.63 | 269.34 | 269.04 | 268.75 | 268.45 | 268.15 | 267.86 | 267.56 |
| Mix Density [g/cc] | 10.96 | 10.91 | 10.85 | 10.80 | 10.74 | 10.69 | 10.63 | 10.58 | 10.52 |
| %U-235 | 2.18 | 2.07 | 1.96 | 1.86 | 1.75 | 1.64 | 1.53 | 1.43 | 1.32 |
| %U-238 | 85.97 | 81.76 | 77.54 | 73.31 | 69.08 | 64.83 | 60.58 | 56.31 | 52.04 |
| %O-16 | 11.82 | 11.83 | 11.85 | 11.86 | 11.87 | 11.88 | 11.90 | 11.91 | 11.92 |
| %O-17 | 0.04 | 0.04 | 0.04 | 0.04 | 0.04 | 0.04 | 0.04 | 0.04 | 0.04 |
| %Th-232 | 0.00 | 4.30 | 8.61 | 12.93 | 17.27 | 21.61 | 25.96 | 30.31 | 34.68 |
| Source: Authors. | | | | | | | | | |

 Table 2: Fuel mix composition in [%wt] to 2.5% of enrichment.

 Table 3: Fuel mix composition in [%wt] to 3.2% of enrichment.

| %ThO2 Comp. | 0 | 5 | 10 | 15 | 20 | 25 | 30 | 35 | 40 |
|-----------------------|--------|--------|--------|--------|--------|--------|--------|--------|--------|
| Total Mass [g] | 269.93 | 269.63 | 269.34 | 269.04 | 268.75 | 268.45 | 268.15 | 267.86 | 267.56 |
| Mix Density [g/cc] | 10.90 | 10.91 | 10.85 | 10.80 | 10.74 | 10.69 | 10.63 | 10.58 | 10.52 |
| %U-235 | 2.79 | 2.65 | 2.51 | 2.38 | 2.24 | 2.10 | 1.96 | 1.82 | 1.69 |
| %U-238 | 85.35 | 81.17 | 76.98 | 72.79 | 68.58 | 64.37 | 60.14 | 55.91 | 51.66 |
| %O-16 | 11.82 | 11.83 | 11.85 | 11.86 | 11.87 | 11.88 | 11.90 | 11.91 | 11.92 |
| %0-17 | 0.04 | 0.04 | 0.04 | 0.04 | 0.04 | 0.04 | 0.04 | 0.04 | 0.04 |

Analyzing Tables 2 and 3 it is observed that the fuel mix compositions in terms of relative percent mass was satisfactorily estimated. Once that the sum of each one relative mass for each one elements and setup it is total in 100.

In order to optimize the results presentation, it was chosen only the fuel compositions that results in k_{inf} values above, approximately, 1.2 once that in the Wichrowski (2017) works it was observed that the criticality just is achieved according to this condition. An evolutive assessment of the k_{inf} evolution according to the rise of the %wtThO2 at the fuel mix it is presented in Table 4.

| k_inf a 2.5% de Enr. | | | k_inf a 3.2 | 9/ wt ThO2 | | | | | |
|----------------------|------------------|-----------|-------------|------------|------------|--|--|--|--|
| | K_inf_ford | K_inf_adj | K_inf_ford | K_inf_adj | 70WU 11102 | | | | |
| | 1.343392 | 1.343375 | 1.403479 | 1.403462 | 0 | | | | |
| | 1.297416 | 1.297401 | 1.360942 | 1.360925 | 5 | | | | |
| | 1.260836 | 1.260820 | 1.320820 | 1.328066 | 10 | | | | |
| | 1225671 | 1.225656 | 1.296750 | 1.296735 | 15 | | | | |
| | 1.190128 | 1.190113 | 1.265090 | 1.265076 | 20 | | | | |
| | 1.153330 | 1.153318 | 1.232198 | 1.232185 | 25 | | | | |
| | | | 1.197450 | 1.197437 | 30 | | | | |
| | Source: Authors. | | | | | | | | |

Table 4: Evolutive analyzes of k_{inf} behavior due to the increase of %wtThO2 at the fuel mix.

In the Table 4, $k_{inf-ford}$ is the one calculated by one of the modules of TSUNAMI and k_{inf_adj} is the, also calculated by one module of TSUNAMI, one used by the General Perturbation Theory to determine the sensitivity of the system.

It should be observed in the Table 4 that are values of k_{inf} highlighted in red, they represent the maximum value, at the level of enrichment, that make possible to achieve the reactor criticality. Based on these values, Tables 5 and 6 present the results for the sensitivity of the fuel mix to the thorium dioxide insertion in the PWR fuel mix. The results in red are for the temperature used as the basis for the study.

| Temp. | %wtThO2 at the Fuel Mix | | | | | | | |
|-----------------|-------------------------|-----------|-----------|-----------|-----------|-----------|--|--|
| [K] | 0 | 5 | 10 | 15 | 20 | 25 | | |
| 280 | -2.04E-02 | -2.87E-02 | -3.06E-02 | -3.05E-02 | -2.89E-02 | -2.61E-02 | | |
| 290 | -2.04E-02 | -2.87E-02 | -3.06E-02 | -3.05E-02 | -2.89E-02 | -2.61E-02 | | |
| 300 | -2.03E-02 | -2.87E-02 | -3.05E-02 | -3.05E-02 | -2.88E-02 | -2.60E-02 | | |
| 310 | -2.02E-02 | -2.85E-02 | -3.04E-02 | -3.03E-02 | -2.87E-02 | -2.59E-02 | | |
| 320 | -2.01E-02 | -2.84E-02 | -3.03E-02 | -3.02E-02 | -2.85E-02 | -2.57E-02 | | |
| 330 | -2.00E-02 | -2.83E-02 | -3.02E-02 | -3,00E-02 | -2.84E-02 | -2.56E-02 | | |
| 340 | -1.99E-02 | -2.82E-02 | -3.00E-02 | -2.99E-02 | -2.82E-02 | -2.54E-02 | | |
| 345 | -1.98E-02 | -2.81E-02 | -3.00E-02 | -2.98E-02 | -2.81E-02 | -2.53E-02 | | |
| Courses Anthony | | | | | | | | |

Table 5: Sensitivity Coefficients to the Fuel Mix at 2.5% of Enrichment.

Source: Authors.

Table 6: Sensitivity Coefficients to the Fuel Mix at 3.2% of Enrichment.

| Temp. | %wtThO2 at the Fuel Mix | | | | | | | | | |
|-------|-------------------------|-----------|-----------|-----------|-----------|-----------|-----------|--|--|--|
| [K] | 0 | 5 | 10 | 15 | 20 | 25 | 30 | | | |
| 280 | -3.49E-02 | -4.34E-02 | -4.57E-02 | -4.60E-02 | -4.47E-02 | -4.23E-02 | -3.92E-02 | | | |
| 290 | -3.49E-02 | -4.34E-02 | -4.57E-02 | -4.60E-02 | -4.47E-02 | -4.23E-02 | -3.92E-02 | | | |
| 300 | -3.49E-02 | -4.34E-02 | -4.56E-02 | -4.59E-02 | -4.46E-02 | -4.23E-02 | -3.91E-02 | | | |
| 310 | -3.48E-02 | -4.33E-02 | -4.55E-02 | -4.58E-02 | -4.45E-02 | -4.21E-02 | -3.90E-02 | | | |
| 320 | -3.47E-02 | -4.32E-02 | -4.54E-02 | -4.57E-02 | -4.44E-02 | -4.20E-02 | -3.89E-02 | | | |
| 330 | -3.46E-02 | -4.31E-02 | -4.53E-02 | -4.56E-02 | -4.43E-02 | -4.19E-02 | -3.87E-02 | | | |
| 340 | -3.46E-02 | -4.30E-02 | -4.52E-02 | -4.55E-02 | -4.42E-02 | -4.18E-02 | -3.89E-02 | | | |
| 345 | -3.45E-02 | -4.30E-02 | -4.52E-02 | -4.55E-02 | -4.41E-02 | -4.17E-02 | -3.85E-02 | | | |
| | Source: Authors | | | | | | | | | |

Source: Authors.

Graphically the behavior due to the fuel sensitivity coefficients can be plotted and studied. Figure 4 and 5 are examples.





Source: Authors.

Figure 5: Behaviour of the sensitivity coeficients for the fuel 3.2% enriched and 25%wtThO2.



Source: Authors.

It is should be noted that the sensitivity coefficients rise up whith the increase of the %wtThO2 in the fuel mix likewise for the increase on the moderator temperature. These facts combined demonstrate that the insertion of ThO2 on the PWR fuel mix increase the intrinsic safety of the nuclear reactor. From Graphs 3 and 4 it is possible to see that the sensitivity coefficients presented a polynomial quadratic behavior, according to the increase in the moderator temperature, confirmed by the correlation coefficient tending to one.

4. CONCLUSION

According to the results obtained by the simulations, it can be verified that the k_{inf} is very sensitive to the variation in the ratio moderator/coolant of a PWR nuclear reactor, at least in the temperature range studied, as was also noticed by Neto (2015). Another point to be observed is that in the evolutive assessment of the behavior of the k_{inf} , due to the increase of ThO₂ at the fuel mix (Table 4), it is possible to confirm that is necessary an increase at the enrichment as the %wtThO₂ on the fuel composition increases, due to the behavior of the thorium, that, initially, acts like an absorber in the process of conversion in ²³³U.

In terms of the safety assessment by the sensitivity analyses, it is shown that the increase of thorium dioxide at the fuel composition of a PWR has not a linear behavior with the sensitivity, once that the sensitivity initially increases with the increase of ThO2 and, after a certain amount of ThO2, it starts to decreases. Although all of the sensitivity coefficients whit the addiction os ThO2 are higher than those fuel mix without thorium insertion. Another point that needs attention is the k_{inf} behavior and values, presented on Table 4, it shows that it is not possible to apply this configuration of fuel since that the k_{inf} is less than 1.2 even in the absence of the absorbers or burnable poisons, like the settings used in the setup of the simulations.

This work shows that it is perfectly possible to use thorium and uranium oxides in a mix as fuel material in a Pressurized Water Reactor.

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